Student Functionality

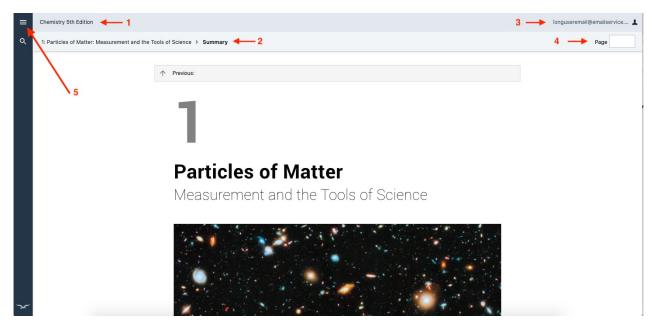
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The Norton Ebook Reader provides an active reading experience, enabling students to take notes, search, highlight, and read offline. This page provides details on how to access and manage these tools.

Hide All Answers

How do I navigate my Ebook?

This section provides details on how users can navigate and search through the new ebook reader.



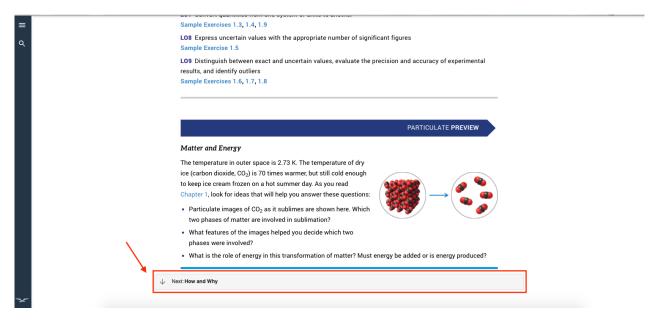
- 1. When logged in you see the **Book Title** displayed at the top of the page.
- 2. The **Chapter and Section** you are working in are visible below the Book Title.
- 3. You can open the **Account Menu** by selecting your username. More information about the features available from this menu are discussed here.
- 4. Indicates what **Page** you are currently viewing.
- 5. Select the three horizontal lines to make the **Table of Contents** appear. More information about this feature can be found here.

How do I navigate between sections?

To view the previous section of the ebook scroll to the top of the page you are currently viewing. Select the box showing the previous section name.

≡	Chemistry 5th Edition	longuseremail@emailservice 💄
۹	1: Particles of Matter: Measurement and the Tools of Science > How and Why	Page
	Previous: 1: Particles of Matter: Measurement and the Tools of Science]
	1.1 How and Why	
~	For thousands of years, humans have sought to better understand the world around us. For most of that time we resorted to mythological explanations of natural phenomena. Many once believed, for example, that the Sun rose in the east and set in the west because it was carried across the sky by a god driving a chariot propelled by winged horses. In recent times we have been able to move beyond such fanciful accounts of natural phenomena to explanations based on observation and scientific reasoning. Unfortunately, this movement toward rational explanations has not always been smooth. Consider, for example, the contributions of a man whom Albert Einstein called the father of modern science, Galileo Galilei. At the dawn of the 17th century, Galileo used advanced telescopes of his own design to observe the movement of the planets and their moons. He concluded that they, like Earth, revolved around the Sun. However, this view conflicted with a belief held by many religious leaders of his time that Earth was the center of the universe. In 1633 a religious tribunal forced Galileo to disavow his conclusion that Earth orbited the Sun and banned him (or anyone) from publishing the results of studies that called into question the Earth-centered view of the universe. The ban was not completely lifted until 1835—nearly 200 years after Galileo's death. In the last century, advances in the design and performance of telescopes have led to the astounding discovery that we live in an expanding universe that probably began 13.8 billion years	
nttps://s	astounding discovery that we live in an expanding universe that probably began 13.8 billion years services-ebook-uat.wwmorton.net/nerd/6806/r/goto/cti/221/4 ago with an enormous release of energy. In this chapter and in later ones, we examine some of the	

To view the next section of the ebook scroll to the bottom of the page you are currently viewing. Select the box indicating the next section.



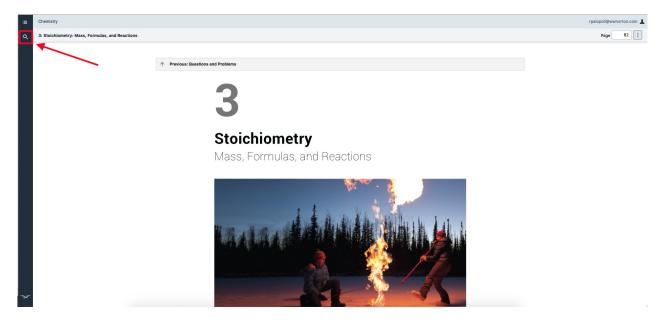
More information about navigating the ebook can be found in the Table of Contents section.

Additionally, you can view your current location by selecting the three dots found underneath the book title.



How do I search the ebook content?

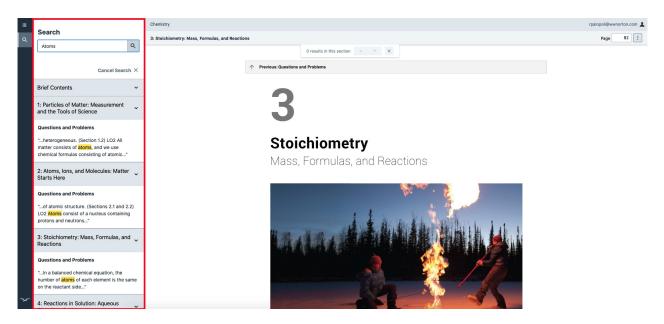
To search the text, select the magnifying glass from the left-hand side of the screen.



Enter a term in the search field.

=		Chemistry	
۹	Search	3: Stoichiometry: Mass, Formulas, and Reactions	Page 82 :
	Atoms Q		
		↑ Previous: Questions and Problems	
	C Type in the field above to search the book	3	
		Stoichiometry	
		Mass, Formulas, and Reactions	

See the full book search results displayed below.



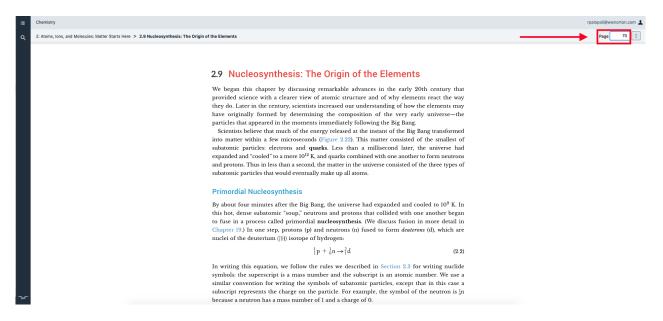
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=		Chemistry rpalopol@wwm	orton.com 💄
٩	Search	1: Particles of Matter: Measurement and the Tools of Science Page	2 🗄
	Atoms Q Cancel Search X	0 of 6 results in this section 🔺 🗸 X	
	cancel search A	ANCIENT UNIVERSE The colors of the more than 10,000 galaxies in this image give us a glimpse into	
	Brief Contents ~	the universe as it existed about 13 billion years ago. This image was taken by NASA's Hubble Space Telescope.	
	1: Particles of Matter: Measurement and the Tools of Science		
	Questions and Problems	PARTICULATE REVIEW	
	"heterogeneous. (Section 1.2) LO2 All matter consists of atoms, and we use chemical formulas consisting of atomic"	Atoms Ind Molecules: What's the Difference? In Chapter I we explore how chemists classify different kinds of matter,	
	2: Atoms, lons, and Molecules: Matter Starts Here	from elements to compounds to mixtures. Hydrogen and helium were the first two elements formed after the universe began. Chemistis use distinctively colored sphere to distinguish attomation of different elements	
	Questions and Problems "of atomic structure. (Sections 2.1 and 2.2) LO2 <mark>Atoms</mark> consist of a nucleus containing protons and neutrons"	in their drawings and models. For example, hydrogen is almost always depicted as white. • How many of the following particles are shown in this image? • Hydrogen itoms?	
	3: Stoichiometry: Mass, Formulas, and Reactions	Hydrogen molecules? Helium atoms?	
	Questions and Problems	 Are molecules composed of atoms, or are atoms composed of molecules? 	
	"In a balanced chemical equation, the number of <mark>atoms</mark> of each element is the same on the reactant side"	SHOW ANSWER	
×	4: Reactions in Solution: Aqueous	Learning Outcomes	

Please Note: These search results are for the entire book. Anytime the keyword you entered is displayed in the text, it will show up here. If you would like to view the help notes on searching the Table of Contents, please click here.

How do I search by page number?

You can search by a specific page number by inserting a number into the page field on the righthand of the screen. This box will display the current page number you are viewing.



After inserting a number in the page and select return on your keyboard, the ebook will take you to that page.

How do I highlight text?

Personal Highlights

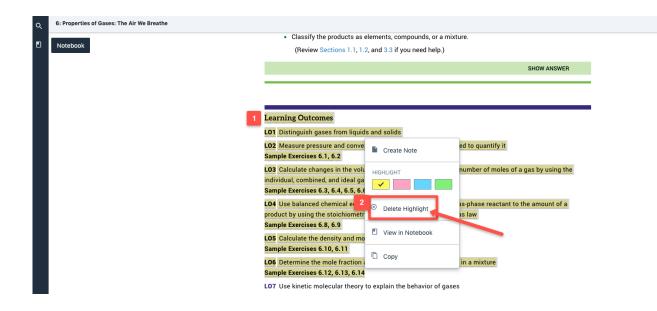
To highlight text, use the cursor to select the text you would like to highlight and the **Context Menu** will appear. Select the color in which you would like the text highlighted: yellow, pink, blue, or green.

1 Learning Outcomes		
LOT Use Avogadro's number and th Sample Exercises 3.1, 3.2, 3.3, 3.4,		lations
LO2 Write balanced chemical equal Sample Exercises 3.8, 3.9, 3.10, 3.1		ctions
LO3 Use balanced chemical equation formed Sample Exercises 3.12, 3.13 LO4 Determine an empirical formul Sample Exercises 3.14, 3.15, 3.16 LO5 Determine a molecular formula	HIGHLIGHT	nt consumed to the mass of a product a substance nolar mass of a substance
Sample Exercise 3.17 LO5 Use data from combustion rea Sample Exercises 3.18, 3.19 LO7 Determine the limiting reactan Sample Exercises 3.20, 3.21	ctions to determine empirical for	rmulas of substances
LOB Calculate the theoretical and p Sample Exercises 3.22, 3.23	ercent yields in a chemical react	tion

Removing Highlights

To remove a highlight, use the cursor to select the text from which you would like to remove the highlighting and the **Context Menu** will appear.

Click Delete Highlight



Select **Delete** and the highlighting will be removed from the selected text.

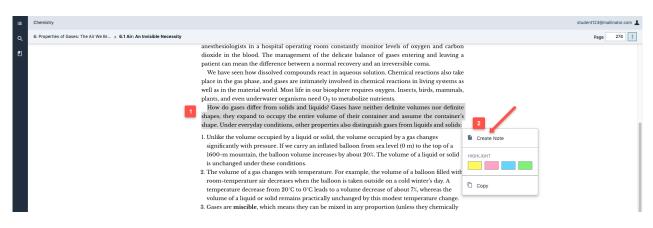
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How do I create annotations?

Personal Annotations

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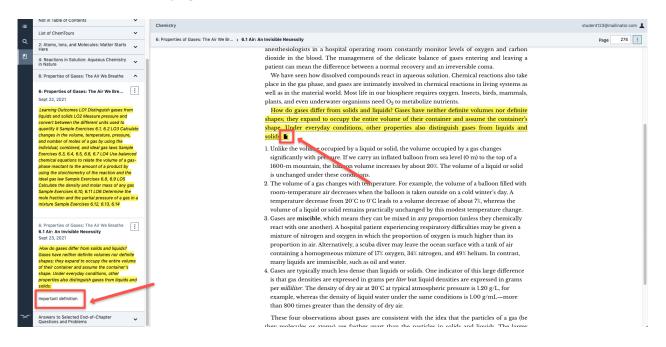
Click Create Note



Type your annotation into the text field and click the **Save** button save your annotation.

	Chemistry	student123@mailinator.com
Create note ×	6: Properties of Gases: The Air We Br > 6.1 Air: An Invisible Necessity	Page 274 :
C: Properties of Gases: — 6.1 Air: An Invisible Necessity Hoard opase differ from solids and liquids? Gases have notified definite volumes nor definite abases: they expand to occupy the entite volume of their container and assume the container and abase. Under everyday conditions, other properties also distinguish pases from liquids and abadd: HIGHLIGHT	anesthesiologists in a hospital operating room constantly monitor levels of oxygen and carbon dioxide in the blood. The management of the delicate balance of gases entering and leaving a patient can mean the difference between a normal recovery and an inreversible coma. We have seen how dissolved compounds react in aqueous solution. Chemical reactions also take place in the gas phase, and gases are intimately involved in chemical reactions in living systems as well as in the material world. Most life in our biosphere requires oxygen. Insects, birds, mammals, plants, and even underwater organisms need 0 ₂ to metabolize nutrients. How do gases differ from solids and liquids? Gases have neither definite volumes nor definite shapes, they expand to occupy the entire volume of their container and assume the container's	
NOTE Important definition	 shape. Under everyday conditions, other properties also distinguish gases from liquids and solids: 1. Unlike the volume occupied by a liquid or solid, the volume occupied by a gas changes significantly with pressure. If we carry an inflated balloon from sea level (0 m) to the top of a 1600-m mountain, the balloon volume increases by about 20%. The volume of a liquid or solid is sunchanged under these conditions. 2. The volume of a gas changes with temperature. For example, the volume of a balloon filled with room-temperature air decreases when the balloon is taken outside on a cold winter's day. A temperature decrease from 20°C to 0°C leads to a volume decrease of about 7%, whereas the volume of a biquid or solid remains practically unchanged by this modest temperature change. 3. Cases are miscible, which means they can be mixed in any proportion (unless they chemically react with one another). A hospital patient experiencing respiratory difficulties may be given a mixture of nitrogen and oxygen in which the proportion of oxygen is much higher than its proportion is in the tark of air. 	

Click on the Notebook page icon to view notes in the Notebook



How to Edit Annotations

Click the notebook page icon. The Context Menu will appear. Select Edit Note

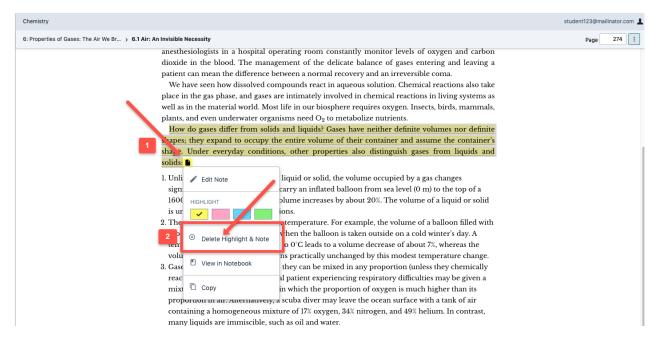
Chemistry			student123@mailinator.com		
Properties of Gases: The Air We Br > 6.1	Properties of Gases: The Air We Br > 6.1 Air: An Invisible Necessity				
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	patient can mean the difference b	between a normal recovery and an irreversible coma.			
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1	well as in the material world. Mo	st life in our biosphere requires oxygen. Insects, birds, mammals,			
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	hape Under everyday condition	ons, other properties also distinguish gases from liquids and			
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	3. Gase 🛄 View in Notebook	they can be mixed in any proportion (unless they chemically			
	reac	al patient experiencing respiratory difficulties may be given a			
	тіхі 🗖 Сору	in which the proportion of oxygen is much higher than its			
	proportion in an enternatively, a scuba diver may leave the ocean surface with a tank of air				
	containing a homogeneous mix	xture of 17% oxygen, 34% nitrogen, and 49% helium. In contrast,			
	many liquids are immiscible, su	uch as oil and water.			
	4. Gases are typically much less d	ense than liquids or solids. One indicator of this large difference			

After editing the note, select **Save**.

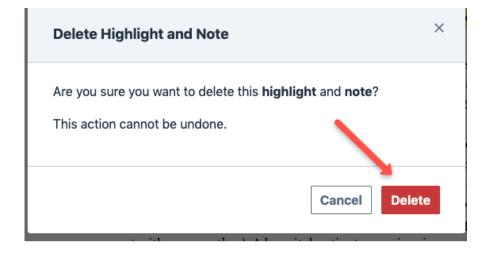
	Chemistry	student123@mailinator.com
Edit note ×	6: Properties of Gases: The Air We Br > 6.1 Air: An Invisible Necessity	Page 274
6: Properties of Gases 8:1 Air: An Invisible Necessity How do gases differ from solids and liquids? Gases have nother definite volumes nor definite shape. Unkey expand to occupy the entire volume of their container and assume the container's shape. Under everyday conditions, other properties also distinguish gases from liquids and solids.	anesthesiologists in a hospital operating room constantly monitor levels of oxygen and carbon dioxide in the blood. The management of the delicate balance of gases entering and leaving a patient can mean the difference between a normal recovery and an irreversible coma. We have seen how dissolved compounds react in aqueous solution. Chemical reactions also take place in the gas phase, and gases are intimately involved in chemical reactions in living systems as well as in the material world. Most life in our biosphere requires oxygen. Insects, birds, mammals, plants, and even underwater organisms need O ₂ to metabolize nutrients. How do gases differ from solids and liquids' Gases have neither definite volumes nor definite	
	shapes: they expand to occupy the entire volume of their container and assume the container's shape. Under everyday conditions, other properties also distinguish gases from liquids and solids.	
NOTE This item will be on the examt The cancel Save	 Unlike the volume occupied by a liquid or solid, the volume occupied by a gas changes significantly with pressure. If we carry an inflated balloon from sea level (0 m) to the top of a 1600-m mountain, the balloon volume increases by about 20%. The volume of a liquid or solid is unchanged under these conditions. The volume of a gas changes with temperature. For example, the volume of a balloon filled with room-temperature air decreases when the balloon is taken outside on a cold winter's day. A temperature decrease from 20 °C to 0 °C leads to a volume decrease of about 7%, whereas the volume of a liquid or solid remains practically unchanged by this modest temperature change. Gases are miscible, which means they can be mixed in any proportion (unless they chemically react with one another). A hospital patient experiencing respiratory difficulties may be given a 	

How to Delete Annotations

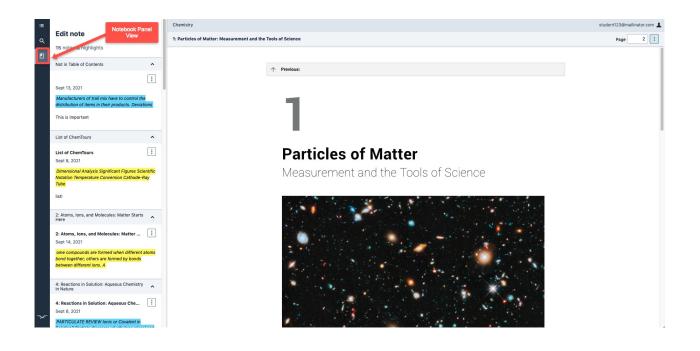
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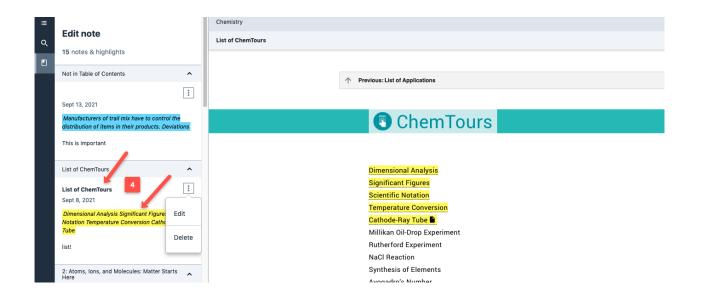


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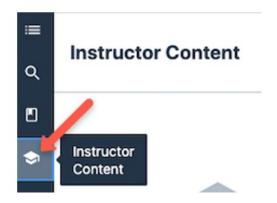
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- 1. This is the **total number** of notes and highlights
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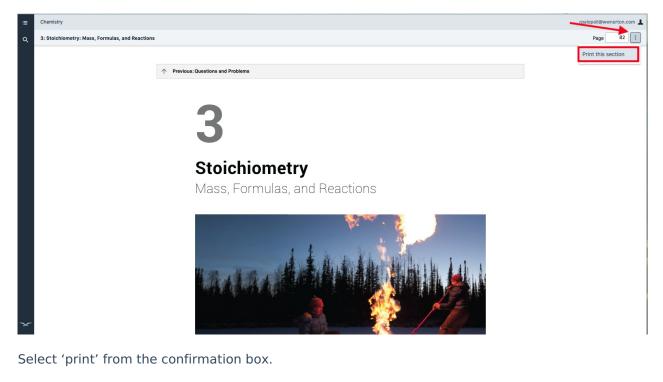
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≔	An Introduction to America's Music						cri@evergr	eener.e	. 1
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			A arena in which the dran the three European empires th Spain, France, and England-bo religion figured differently in e that served religion differed an thus begins by considering the Spanish and British, brought to change in a new cultural settin	hat dominated the so oth commerce and each of those empiri- mong them as well. Is sacred music that to the New World and	ettlement of North Ameri religion played key roles. res, and consequently the Our study of America's m two of those empires, the	ca– But music usic			
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Soul": Secular Music in the Colonies and Early Republic	> Chapter 2: "The Favorite Passion of My Soul": Secular Music in the Colo > Song, Dance, And	Home Music Making	Page 41 🗄
Song, Dance, And Home Music Making			
Spotlight on History: Francis Hopkinson, Music Amateur	Previous: Chapter 2: "The Favorite Passion of My Soul": Secular Mus	ic in the Coloni	
Military, Concert, and Theater Music	Print ×		
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	Cancel Print uring the 17 a few cities on the Eastern Seaboard–Boston, New York, Philadelphia, Baltimore, and Charleston–where immigrant	ntrated in	
	musicians practiced what they had learned in Europe. Tha points to something basic in America's musical life from ti of European settlement, musicians here have been able to granted the ample supply of music from the British Isles a European Continent, made available through oral traditior	at fact he start o take for nd the n and	
	written notation. Given a steady supply from Europe, then little demand outside religious circles for music by Americ compared in the according to according the steady of the second	can	

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≡	Chapter 2: "The Favorite Passion of My	An Introduction to America's Music Isimp:	onsrcri@everg	greener.e 👤
Soul": Secular Music in the Colonies and Early Republic		> Chapter 2: "The Favorite Passion of My Soul": Secular Music in the Colonie > Military, Concert, and Theater Music	c Page	50 :
	Song, Dance, And Home Music Making	and out of battle. Military life relies a good deal on ceremonies. Parades that feature uniformed soldiers marching in time to music display		
	Spotlight on History: Francis Hopkinson, Music Amateur	discipline, suggest invincibility, and contribute to an army's goal of deterring wars as well as fighting them. Finally, a military unit is both a fighting force and a society, and		
	Military, Concert, and Theater Music	its musicians have long performed at concerts, mealtime performances, evening entertainments, sports festivals, and riding		
	Questions for Discussion and Review	Print illitary in the 1700s A functional and the d music-which		
		Copyright Restriction You have reached your printing limit. You can no longer print pages from this eBook. ises translated as ie made up of pairs ccasionally flutes or they were hired by n pockets.		
		10 of 10 section printed 0 sections remaining gesprit de corps, remonies. The band		
		nonized sound ation, though it could also be an inspirational and ceremonial presence. When colonists formed militia units to fight in the American Revolution, they followed British custom, including the two-part division into field music and Harmoniemusik. Around 1793 Philip Phile, an immigrant musician who worked in American theaters, composed the most enduring American instrumental composition of the 1700s, The President's March (LG 2.3). In 1798 a Philadelphia judge, Joseph Hopkinson, set patriotic words to the march, beginning "Hail Columbia, happy land," and the combination of tune and text held a place as a favorite national song, "Hail Columbia, well into the twentieth century. The tuneful melody and sprightly march rhythm combine to create a piece of		

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۹	Part 1: American Colonies to 1763 \rightarrow Chapter 1: A New World	Page 48
	Part 1: American Colonies to 1783 Chapter 1: A New World Image: the second s	Page 48 2
	contact with peoples on other continents, the inhabitants of North and South America had developed no immunity to the germs that also accompanied the colonizers. As a result, they suffered a series of devastating epidemics, the greatest population catastrophe in human	

You may also see a message that indicates only part of the section is available for printing.

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Part 1: American Colonies to 1763> Chapter 1: A New World	Page 48
Previous: The First Americans The discovery of America, [*] the British writer Adam Smith announced in his celebrated work The We [*] Print	
	Per 1: American Colonies to 173 Chapter 1: A New World Previous: The First Americans The discovery of America," the British writer Adam Smith announced in his celebrated work The We Print Print Do you want to print this section? Nou are allowed to print up to <#> sections. Indian Nou are allowed to print up to <#> sections. Indian Nou are allowed to print due to copyright restriction It had a of e. The istence, daniele intersee inter

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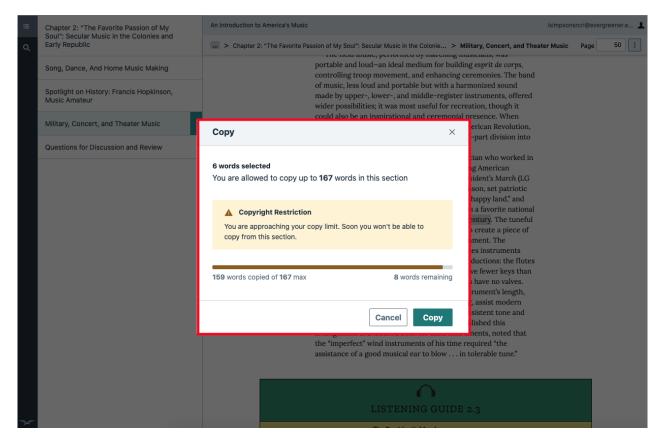
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Q	4: Reactions in Solution: Aqueous Chemistry in Nature	> 4.1 Ions and Molecules in Oceans and Cells	Page 144 🗄
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		ponds, ne to arise. Cancel Copy of ions for cells biological systems non-on-carry name or carry name in province me metium in which dissolved molecules and ions moved about, collided with one another, and participated in chemical reactions that led to the formation of more complex molecules, and eventually to assemblies of molecules capable of reproduction.	
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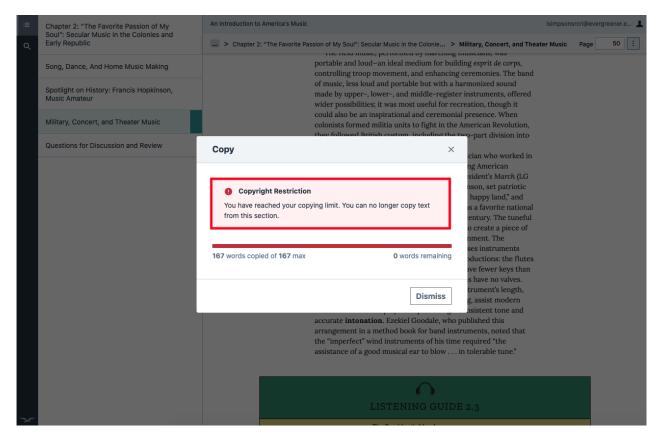
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	Soul": Secular Music in the Colonies and Early Republic	> Chapter 2: "The Favorite Passion of My Soul": Secular Music in the Colonie		ter Music	Page	50
	Song, Dance, And Home Music Making	portable and loud–an ideal medium for b controlling troop movement, and enhance	uilding esprit de corps, ing ceremonies. The band			
	Spotlight on History: Francis Hopkinson, Music Amateur	of music, less loud and portable but with made by upper-, lower-, and middle-regi wider possibilities; it was most useful for	ster instruments, offered recreation, though it			
	Military, Concert, and Theater Music	Сору	 resence. When herican Revolution, part division into 			
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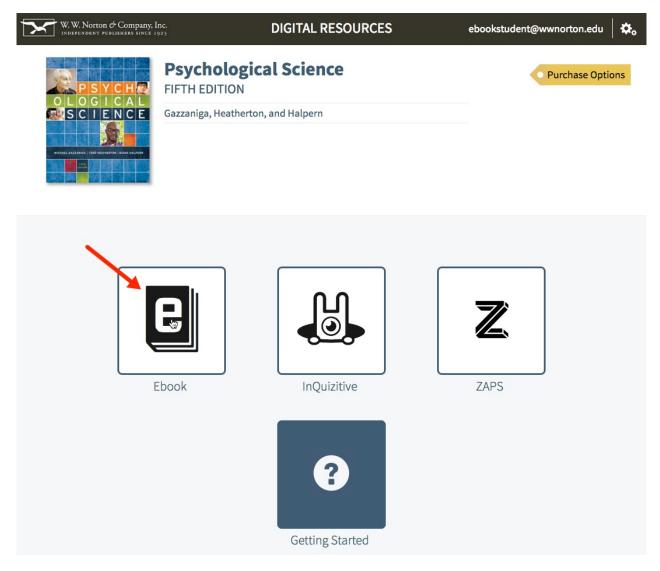
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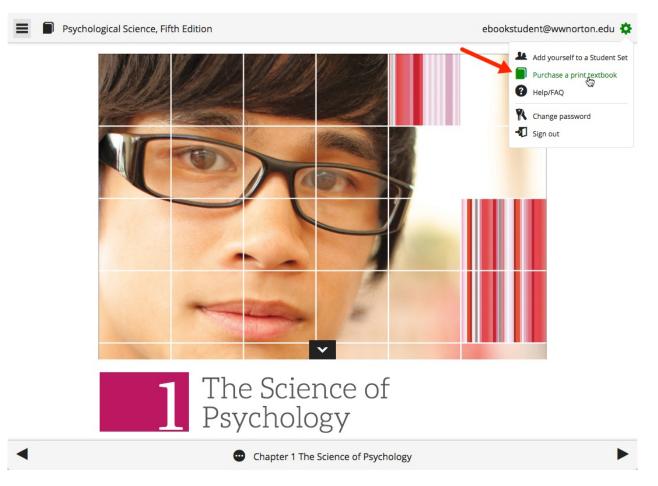
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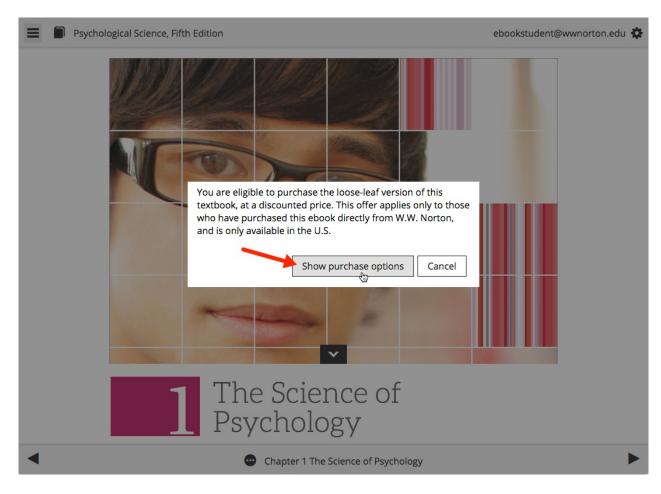
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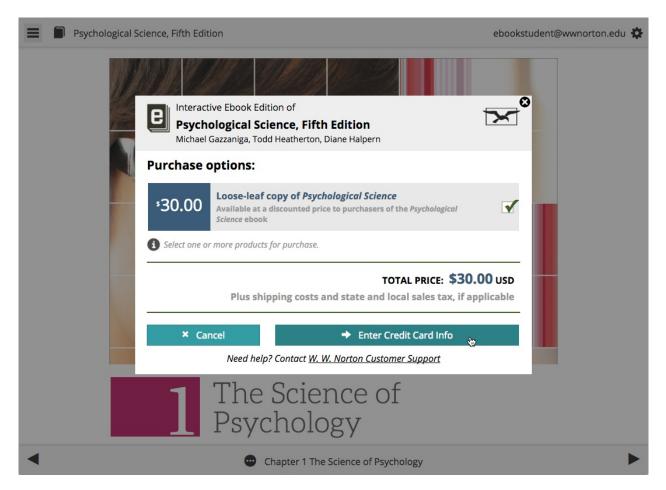
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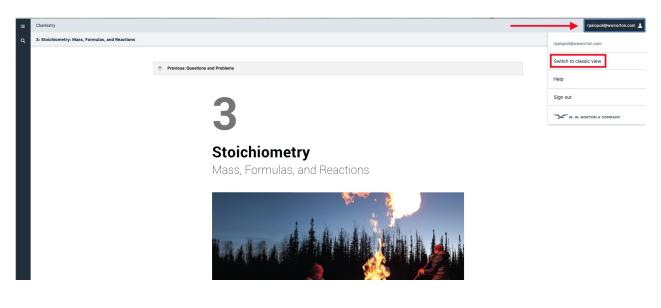


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